

Chemomicin A, a New Angucyclinone Antibiotic Produced by *Nocardia mediterranei* subsp. *kanglensis* 1747-64

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Abstract A new angucyclinone antibiotic, chemomicin A was isolated from cultured broth of *Nocardia mediterranei* subsp. *kanglensis* 1747-64. Its chemical structure was determined to be 1,2,3,4a,5,6,6a,12a,12b-nonahydro-1,2,3,8,12,12b-hexahydroxy-3-methyl-6a,12a-epoxybenz-[a]anthracen-4,7(12H)-dione by a detailed spectroscopic analysis. Chemomicin A had antimicrobial activity against *Bacillus subtilis* and *Enterococcus faecium* with MIC values of 10.2 and 20.4 μM , respectively, and showed cytotoxicity against human colorectal cancer HCT116 cells and human esophageal carcinoma YES-2 cells with IC_{50} values of 127 and 153 μM , respectively.

Keywords angucyclinone, *Nocardia mediterranei* subsp. *kanglensis* 1747-64, chemomicin A, SF2315B, structural elucidation

Nocardia mediterranei subsp. *kanglensis* 1747-64 [1] (strain 1747-64), an angucyclinones producing strain, was isolated from soil sample collected from Kang-Le Area, Guangdong Province, P. R. China. Two new angucyclinones, kanglemycin C (**2**) [2] and kanglemycin M (**3**) [3] (Fig. 1), were found in the cultured broth of the strain. Aromatic protons (H-9, H-10, H-11) in D-ring of most angucyclinones [4], like PD116779 [5], rubiginones [6], EI-1507-1, EI-1507-2 [7] and ochracenomicins A, B and C

[8], showed ABX coupling system in the downfield region (δ 7~8) in their $^1\text{H-NMR}$ spectra. Based on the hypothesis that the strain was a talented producer of angucyclinones like rubiginones producing strain [6], a project to use spectroscopic data mentioned above as a probe in HPLC-NMR to further screen new angucyclinones from the cultured broth of the strain 1747-64 was carried out. As a result, new members of angucyclinone group designated as chemomicins were early identified by HPLC-NMR. Among them, chemomicin A (**1**) was further isolated under the guidance of its retention time in HPLC. Structural studies showed **1** was a unique angucyclinone with six hydroxyl groups, a 4-carbonyl group and a 6a,12a-epoxide functional group (Fig. 1). In this paper, we wish to report the fermentation, isolation, physico-chemical properties, structure elucidation and biological activities of **1**.

A stock culture of the strain 1747-64 was maintained on Gause No. 1 agar slant consisting of KNO_3 0.1%, NaCl 0.05%, K_2HPO_4 0.05%, $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$ 0.001%, $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ 0.05%, soluble starch (Beijing Qi Te Xin Chemical Co., Ltd., China) 2.0%, and agar 1.5% (pH 7.0) at 4°C. The stock culture was transferred into 250-ml Erlenmeyer flasks containing 50 ml of seed medium consisting of glucose 3.0%, yeast extract (Shanghai Yeast Manufactory, China) 0.5%, $(\text{NH}_4)_2\text{SO}_4$ 0.5% and CaCO_3 0.5% (pH 6.5). The culture was incubated on a rotary shaker (220 rpm) at 28°C for 48 hours. Five milliliters of the seed culture was transferred to 500-ml Erlenmeyer flasks containing 100 ml

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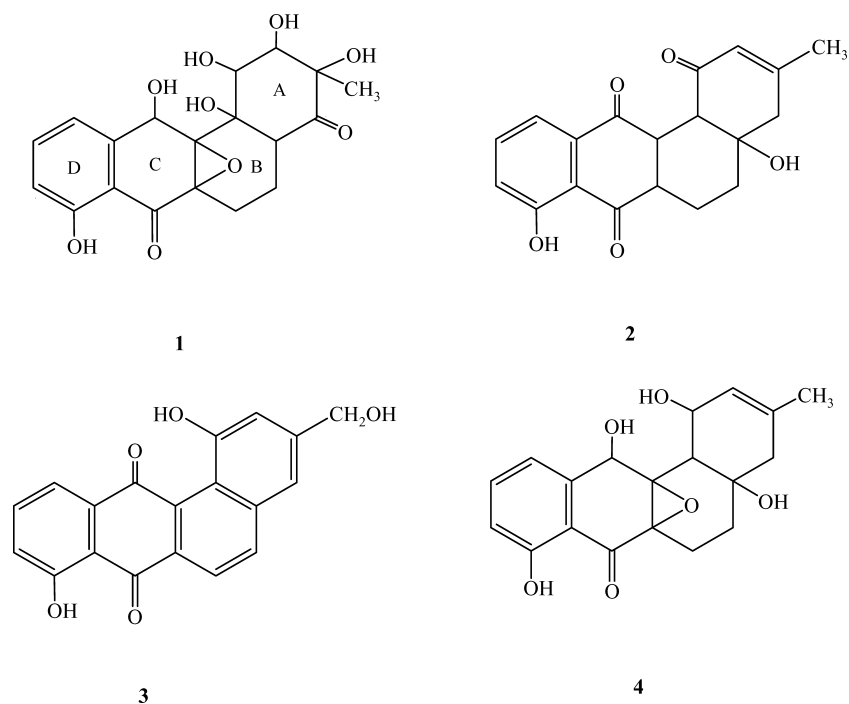


Fig. 1 Structures of chemomycin A (**1**), kanglemycin C (**2**), kanglemycin M (**3**) and SF2315B (**4**).

of the producing medium consisting of glucose 4.0%, yeast extract 1.0%, peanut meal 0.5%, peptone (Shanghai Donghai Pharmaceutical Manufactory, China) 0.5% and CaCO_3 0.1% (pH 6.5). The fermentation was carried out at 28°C for 96 hours on a rotary shaker (220 rpm).

The fermentation broth (25 liters) was adjusted to pH 5.0 with 2 N HCl and filtered. The filtrate was extracted with ethyl acetate (25 liters). The extract was concentrated to a small volume under reduced pressure to give a syrup. It was then chromatographed on a column of silica gel (300 ml, Qindao Silica Manufactory, China) and developed with CHCl_3 -MeOH, 19:1 (v/v 1500 ml). Thirty fractions (50 ml per fraction) were collected and 200 μl solutions from each fraction were dried in 1.5 ml eppendroff tubes with N_2 stream. Then, the resulting residues were dissolved in 50 μl methanol for analysis with HPLC on a Zorbax SB-C18 column (9.4 \times 250 mm, 5 μm , Agilent), with MeOH- H_2O , 65:35 (v/v) at 1 ml/minute after filtered through 0.22 μm membrane. Fractions (No. 21 to 23) containing most of **1** were pooled to yield sample (166 mg). The sample (83 mg/ml, 2 ml) was further purified by HPLC on a shim-pack PRC-ODS column (250 \times 20 mm, Shimadzu) with MeOH- H_2O , 55:45 (v/v) at 4 ml/minute to yield 40 mg of **1** as white powder.

The physico-chemical properties of **1** are summarized in Table 1. The molecular formula of **1** was established as $\text{C}_{19}\text{H}_{20}\text{O}_9$, on the basis of high-resolution SI-MS, NMR

spectra in CD_3OD (I) and in $\text{DMSO}-d_6$ (II). The $^1\text{H-NMR}$ and $^{13}\text{C-NMR}$ spectral data of **1** are shown in Table 2.

Analysis of $^1\text{H-NMR}$ and $^{13}\text{C-NMR}$ together with DEPT and a heteronuclear single quantum coherence (HSQC) indicated 19 carbon signals of **1** could be attributed to two carbonyl carbons, six aromatic carbons (including three methine groups, two quaternary carbons, one oxygen-substituted quaternary carbon), seven other oxygen-substituted carbon signals (including three oxygenated methine groups and four oxygenated quaternary carbons), one methine group, two methylene groups, one methyl group. Six hydroxyl protons at δ 11.4 (H-bonded phenolic hydroxyl proton), 7.1, 6.9, 6.5, 5.4, and 5.1 appeared in $^1\text{H-NMR}$ spectrum in $\text{DMSO}-d_6$, but disappeared in CD_3OD .

Epoxide functional group existed in **1** was early deduced from calculating the oxygen atom number and the number of oxygen-substituted carbon signals, since only six oxygen atoms have to be assigned to seven oxygen-substituted carbon signals except that three oxygen atoms were ascribed to two carbonyl groups and one phenolic hydroxyl group.

The IR spectrum indicated the presence of hydroxyl groups (3410 cm^{-1}), a ketone carbonyl group (1714 cm^{-1}) and a chelated carbonyl group (1653 cm^{-1}), which were further confirmed by a ketone carbonyl carbon signal at δ 204.02 (C-4) and a chelated carbonyl carbon signal at δ 201.20 (C-7) in $^{13}\text{C-NMR}$ spectrum in $\text{DMSO}-d_6$, the same

Table 1 Physico-chemical properties of **1**

| | |
|---|--|
| Appearance | White powder |
| Molecular weight | 392 |
| Molecular formula | C ₁₉ H ₂₀ O ₉ |
| HRSI-MS (<i>m/z</i>) Found: | 391.1033 (M-H) ⁻ |
| Calcd: | 391.1034 |
| UV λ _{max} ^{MeOH} nm (<i>ε</i>) | 217 (19,282), 260 (8,991), 335 (4,564) |
| IR ν _{max} (KBr) cm ⁻¹ | 3410, 1714, 1653, 1616, 1456, 1250, 1053 |
| Solubility | MeOH, DMSO, CHCl ₃ |
| TLC, Rf value ^a | 0.24 |
| HPLC, Rt (min) ^b | 17.6 |

^a Silica gel 60 F254 (Merck), CHCl₃-MeOH, 19:1 (v/v).

^b Zorbax SB-C18 (9.4×250 mm, 5 μm, Agilent), 65% MeOH, 1 ml/minute, 254 nm.

Table 2 NMR data of **1** in CD₃OD (I) and in DMSO-*d*₆ (II)

| Position | I | | II | |
|-------------------|-----------------------------|--|-----------------------------|--|
| | δ _C ^a | δ _H ^b (mult, J Hz) | δ _C ^a | δ _H ^b (mult, J Hz) |
| 1 | 66.52 | 4.5 (br s) | 64.52 | 4.5 (br s) |
| 1-OH | | | | 6.9 (br s) ^c |
| 2 | 64.77 | 3.5 (d, 2) | 62.45 | 3.5 (d, 2.5) |
| 2-OH | | | | 5.1 (s) ^c |
| 3 | 60.55 | | 58.65 | |
| 3-OH | | | | 7.1 (br s) ^c |
| 3-CH ₃ | 15.16 | 1.4 (s) | 15.29 | 1.3 (s) |
| 4 | 205.15 | | 204.02 | |
| 4a | 50.75 | 2.7 (dd, 12, 3.5) | 49.16 | 2.7 (dd, 12, 3.5) |
| 5 | 16.37 | α 1.8 (m) β 2.1 (m) | 14.89 | α 1.7 (m) β 1.9 (m) |
| 6 | 26.87 | α 1.9 (ddd, 14.5, ~14.5 ^d , 4.5) β 2.1 (m) | 25.33 | α 1.8 (ddd, 14.0, ~14.0 ^d , 4.0) β 2.0 (m) |
| 6a | 78.74 | | 77.22 | |
| 7 | 202.22 | | 201.20 | |
| 7a | 114.81 | | 113.42 | |
| 8 | 162.25 | | 160.16 | |
| 8-OH | | | | 11.4 (s) |
| 9 | 116.53 | 6.8 (d, 8) | 115.15 | 6.8 (d, 8) |
| 10 | 137.45 | 7.5 (dd, 8, 8) | 136.37 | 7.6 (dd, 8, 8) |
| 11 | 121.48 | 7.2 (d, 8) | 120.09 | 7.2 (d, 8) |
| 11a | 145.31 | | 144.76 | |
| 12 | 65.97 | 5.5 (s) | 64.37 | 5.4 (s) |
| 12-OH | | | | 5.4 (s) |
| 12a | 76.32 | | 75.60 | |
| 12b | 80.07 | | 78.41 | |
| 12b-OH | | | | 6.5 (s) |

^a The ¹³C-NMR was measured at 125 MHz. ^b The ¹H-NMR was measured at 500 MHz. ^c Hydroxyl proton assignment may be interchangeable. ^d ³J_{HH} coupling constant between two vicinal protons: Hα-5 and Hα-6 was unresolved and was deduced from multiplicity of signal for Hα-6, which was similar as triplet doublet.

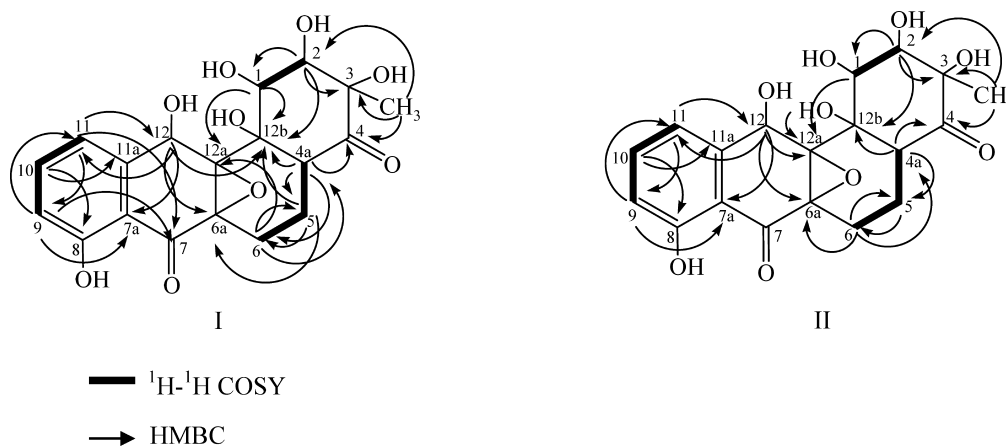


Fig. 2 Summary of ^1H - ^1H COSY and HMBC experiments of **1** in CD_3OD (I) and in $\text{DMSO}-d_6$ (II).

carbon signals could be observed at δ 205.15 and δ 202.22 in CD_3OD . An ABX coupling system of three aromatic proton signals (H-9 to H-11), the probe mentioned above, was readily observed in ^1H -NMR. By tracing the cross peaks from H-9, H-10 and H-11 in HMBC in $\text{DMSO}-d_6$ and in CD_3OD , as shown in Fig. 2, C-7a, C-8, C-11a, and C-12 were assigned. The chemical shifts of C-8 at 160.16 in $\text{DMSO}-d_6$ and at 162.25 in CD_3OD suggested that phenolic hydroxyl proton at δ 11.4 in ^1H -NMR ($\text{DMSO}-d_6$) should be attached to C-8, the peri position of the chelated carbonyl carbon (C-7). It was further supported by HMBC (CD_3OD), in which, both 9-H and 11-H were long-range correlated with C-7. The chemical shifts of C-12 and H-12 together with the data mentioned above indicated the chromophore of **1** is an isosclerone moiety, the same with SF2315B (**4**) [9], as shown in Fig. 1, which has no absorption peak at wavelength longer than 400 nm, usually shown by quinone chromophore [10]. Hydroxyl proton at δ 5.68 (12-OH) of **4** suggested one of two protons at δ 5.4 of **1** should be attributed to hydroxyl proton (12-OH).

H-12 was correlated with C-12a and C-6a in HMBC ($\text{DMSO}-d_6$) and cross peak between H-12 and C-6a could be observed as well in HMBC (CD_3OD). Thus, C-6a and C-12a were assigned. Considering their chemical shift and the reason oxygen atoms were unproportionate with oxygen-substituted carbon signals, as mentioned above, C-6a and C-12a should form an epoxide functional group. The substructure (rings C and D) of **1** was established as shown in Fig. 1 and was further supported by NMR data comparison between **1** and **4** and unambiguous elucidation of another substructure (rings A and B) of **1**.

In another substructure, the two structural fragments: $^6\text{CH}_2$ - $^5\text{CH}_2$ - ^4aCH and $^{12\text{bC}}$ (OH)- ^1CH (OH)- ^2CH (OH)- (CH_3) ^3C (OH)- ^4C (O) were readily identified by ^1H - ^1H

COSY and HMBC. Linkage between the two fragments was established by HMBC. The methine proton at δ 2.7 (H-4a) was coupled to ketone carbonyl carbon at δ 204.02 (C-4) and oxygenated quaternary carbon at δ 78.41 (C-12b) in HMBC ($\text{DMSO}-d_6$), meanwhile, methylene proton at δ 2.1 (5-H) and methine proton at δ 4.5 (1-H) were long range coupled to C-12b in HMBC (CD_3OD). The data above revealed the linkage of two structural fragments through C-4a with C-4 and C-12b.

Long range couplings observed between H-12 and C-12b; between H-6 and C-12a; between H-5 and C-6a in HMBC (CD_3OD) fused two substructures through C6-C6a and C12a-C12b, which were further supported by cross peaks between H-1 and C-12a; between H-6 and C-6a in HMBC ($\text{DMSO}-d_6$) as complementary evidence. Hydroxyl proton at δ 6.5 was assigned to 12b-OH by tracing cross peak from C-12a observed in HMBC ($\text{DMSO}-d_6$). Finally, the planar structure of **1** was established as 1,2,3,4a,5,6,6a,12a,12b-nonahydro-1,2,3,8,12,12b-hexahydroxy-3-methyl-6a,12a-epoxybenz[a]anthracen-4,7(12H)-dione.

In the ^1H -NMR (CD_3OD and $\text{DMSO}-d_6$) of **1**, coupling constants (2 and 2.5 Hz) between H-1 and H-2 showed the vicinal protons were *cis* configuration. Two coupling constants (12 and 3.5 Hz) between H-4a and H-5 α ; H-4a and H-5 β revealed that H-4a and H-5 α were axial protons in *trans* configuration and H-5 β was an equatorial proton. The large $^3J_{\text{HH}}$ coupling constant (\sim 14.5 Hz in CD_3OD and 14.0 Hz in $\text{DMSO}-d_6$) between two vicinal protons: H α -5 and H α -6 indicated anti-periplanar conformation of these protons and chair conformation of the B-ring [11]. In NOE experiments, irradiation of H-2 enhanced obviously the intensity of H-1 and 3- CH_3 , suggesting that H-2 and 3- CH_3 should be *cis* configuration (data not shown). The relative

configuration of **1** remained to be studied in detail.

Within our knowledge, angucyclinones, including **4**, EI-1507-1 and EI-1507-2, rubiginone I, angucyclinone D and elmycin C [4], possess a 6a,12a-epoxide functional group. Most of them were weakly active against Gram-positive and Gram-negative bacteria. For example, EI-1507-1 had weak antimicrobial activities against *Enterococcus faecium*, *Bacillus subtilis* and *Proteus vulgaris* with MIC values of 120, 60 and 60 μM , respectively. On the other hand, all of them have different bioactivities. For example, **4** showed inhibitory activity against reverse transcriptase of avian myeloblastosis virus [12], and both EI-1507-1 and EI-1507-2 inhibited mature interleukin-1 β secretion from THP-1 cell with IC₅₀ values of 1.1 and 1.4 μM , respectively. Preliminary bioactive studies demonstrated **1** had moderate antimicrobial activities against *B. subtilis* and *E. faecium* with MIC values of 10.2 and 20.4 μM , respectively, and showed cytotoxicity against human colorectal cancer HCT116 cells and human esophageal carcinoma YES-2 cells with IC₅₀ values of 127 and 153 μM , respectively. Further studies on detailed biological activities of **1** are in progress.

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